



A New Approach to the Synthesis of the 17B-Butenolide Fragment of Cardenolides

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Abstract: A new, efficient synthesis of the 17β-butenolide fragment characteristic of cardenolides is effected by [2 + 2]-cycloaddition of dichloroketene to 3β-acetoxypregna-5,20-diene, as a key step.

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Cardenolides are plant steroids, which occur as glycosides and possess powerful cardiotonic activity. They are considered as "the most ingested drugs in medicine". Since the pioneering studies by Ruzicka and the first synthesis of digitoxigenin in 1962 enormous efforts have been directed toward synthetic cardenolides. Despite the long history of research in the area, the interest in cardenolides continues. Recently reported work on cardenolides concerns the total synthesis, the biosynthesis, and the search for new, less toxic digitalis-like compounds for therapeutic use with better pharmacological properties. The new methods of introduction of the 17β -butenolide moiety and synthetic approaches to complex cardenolides have also been reported. Besides the 14β -hydroxyl group, the 17β -butenolide moiety is one of the crucial features of cardenolides indispensable for their biological activity. 17-Oxoandrostanes and 21-hydroxy-20-oxopregnanes have been widely used as substrates in the syntheses of cardenolides.

This work is a continuation of our interest in steroidal cyclobutanones. Since the transformation of pregnane derivatives into 17β -butenolide steroids requires a two carbon side chain elongation, the reaction of the appropriate olefinic substrate with a reactive ketene appeared to be an attractive approach to the four carbon side-chain moiety characteristic of cardenolides.

The Scheme 1 illustrates the synthetic pathway. The starting olefin 1, the 3β-acetoxypregna-5,20-diene, was obtained from 3β-acetoxypregn-5-en-20-one following the reported procedure.¹⁰ The regioselective [2 + 2] cycloaddition of 1 and dichloroketene^{11a} afforded dichlorocyclobutanone 2 in 58% isolated yield.^{11b} This could be effectively reduced with zinc in AcOH to 3 or 4, depending on the reaction conditions.¹² However, when the crude cycloaddition product was immediately reduced 4 was isolated in 82% yield.^{11c} At this stage of the synthesis, the 3β-hydroxyl group had to be protected as a TBDMS-ether in two steps: hydrolysis of 4 (K₂CO₃, MeOH; 98% yield of 5) followed by the reaction with t-butyldimethylsilyl chloride (imidazole, DMF, 1h, r.t.) gave 6 in 92% yield. The Baeyer-Villiger oxidation of 6 (30% H₂O₂, MeOH-THF, NaOH) resulted in formation of the lactone 7 (87% yield after short column chromatography) as a 1:1 mixture of C-20 epimers.^{11d} The dehydrogenation of the lactone 7 was achieved by taking advantage of the phenylselenylation-oxidation procedure.¹³ Compound 8 was isolated in 75% yield from the reaction of 7 with LDA and PhSeCl (THF, -70 °C), while oxidation of 8 (30% H₂O₂, THF-AcOH) gave butenolide 9 in 67% yield. The deprotection of the TBDMS-ether afforded the known 3β-hydroxy derivative 10.¹⁴

This method of constructing the butenolide fragment of cardenolides is relatively simple and efficient (the total yield of the five step synthesis of 9 from the readily available 1 is 32 %). The transformation of 14α -card-20(22)-enolide to the Δ^{14} olefin and 14β -hydroxy derivatives has been reported. 15

References and notes

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- (a) Cl₂C=C=O was generated in situ from Cl₃COCl and Zn in Et₂O, under sonification conditions; (b) purification of the crude reaction product on SiO₂ column is usually accompanied by the slow decomposition of the dichlorcyclobutanones (ref. 9b); (c) Δ⁵-double bond was unreactive toward dichloroketene; (d) the ratio was estimated from the integration of the low field signals at δ: 4.47, 4.37, 3.93 and 3.83 in the ¹H NMR spectrum.
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- 16. 2: IR (CHCl₃): v = 1805, 1725 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 5.38$ (br d, J = 4.9 Hz, 1H, 6-H), 4.60 (m, 1H, 3 α -H), 3.22 3.03 (m, 2H, CH₂CO), 2.03 (s, 3H, CH₃CO₂), 1.03 and 1.02 (s, 3H, 19H,), 0.78 and 0.72 (s, 3H, 18-H). 4: m.p. 130-132°C (MeOH),); IR: v = 1775, 1725 cm⁻¹; ¹H NMR (CDCl₃): $\delta = 5.38$ (br d, J = 5.1 Hz, 1H, 6-H), 4.60 (m, 1H, 3 α -H), 3.14 3.00 (m, 2H, CH₂CO), 2.87 2.70 (m, 2H, CH₂CO), 2.03 (s, 3H, CH₃CO₂), 1.03 (s, 3H, 19-H), 0.70 (s, 3H, 18-H). 7: m.p. 201-205 °C; IR (CHCl₃): v = 1775 cm⁻¹; ¹H NMR (CDCl₃): $\delta = 5.33$ (bd, J = 5.2 Hz, 1H, 6-H), 4.47 (dd, J = 8.24 Hz, J = 8.52 Hz, 1H, CH₂O, isomer A) and 4.37 (dd, J = 8.24 Hz, J = 7.97 Hz, 1H, CH₂O, isomer A), 3.93 (dd, J = 9.07 Hz, J = 9.34 Hz, 1H, CH₂O, isomer B) and 3.83 (dd, J = 9.07 Hz, J = 9.61 Hz, 1H, CH₂O, isomer B), 3.48 (m, 1H, 3 α -H), 2.65-2.48 (m, 2H, CH₂CO), 1.00 (s, 3H, 19-H), 0.89 [9H, s, C(CH₃)₃], 0.70 and 0.69 (s, 3H, 18-H), 0.06 [s, 6H, Si(CH₃)₂]. 9: mp: 183-185 °C (heptane), [$\alpha J = -40^{\circ}$ (c = 0.25, CHCl₃); IR (CHCl₃): v = 1785, 1750, 1630 cm⁻¹; ¹H NMR (CDCl₃): $\delta = 5.85$ (d, J = 1.6 Hz, 1H, 22-H), 5.32 (bd, J = 5.2 Hz, 1H, 6-H), 4.83 and 4.69 (ABX, J = 17.6 Hz and 1.6 Hz, 2H, CH₂O), 3.48 (m, 1H, 3 α -H), 1.00 (s, 3H, 19-H), 0.89 [s, 9H, C(CH₃)₃], 0.64 (s, 3H, 18-H), 0.06 [s, 6H, Si(CH₃)₂].